

## Bis-BBN (9-borabicyclo[3.3.1]nonane) adduct of bis(diethylglyoximato)nickel(II)

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### Key indicators

Single-crystal X-ray study  
 $T = 150\text{ K}$   
Mean  $\sigma(\text{C}-\text{C}) = 0.002\text{ \AA}$   
 $R$  factor = 0.025  
 $wR$  factor = 0.031  
Data-to-parameter ratio = 16.5

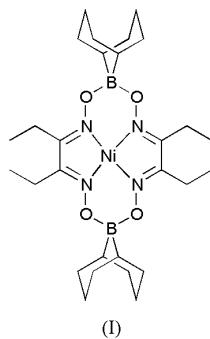
For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The bis-BBN adduct of bis(diethylglyoximato)nickel(II), [2,9-bis(cyclooctane-1,5-diy)-4,6,11,13-tetraethyl-1,3,8,10-tetraoxa-4,7,11,14-tetraaza-2,9-diboracyclotetradecane-4,6,11,13-tetraene- $\kappa^4\text{N}$ ]nickel(II),  $[\text{Ni}(\text{C}_{28}\text{H}_{48}\text{B}_2\text{N}_4\text{O}_4)]$ , crystallizes as a monomer, with no short  $\text{Ni}\cdots\text{Ni}$  contacts. The asymmetric unit contains only half a molecule and the Ni atom lies on an inversion centre.

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### Comment

In the preceding paper, we reported the structure of a mono-9-borabicyclo[3.3.1]nonane (BBN) adduct of bis(diethylglyoximato)nickel(II) (Krivokapić *et al.*, 2003). We report here the structure of the bis-adduct, (I). This compound was prepared in good yield by treating bis(diethylglyoximato)nickel(II) with methoxy-9-BBN in toluene at reflux for 2 d.



The geometry of (I) (Fig. 1 and Table 1) is similar to those of previously reported complexes of this type (Chakravorty, 1974; Krivokapić *et al.*, 2003). Compound (I) adopts a  $C_i$  conformation, with approximate  $C_{2h}$  symmetry, with the BBN units shifted towards opposite faces of the macrocycle; this conformation evidently prevents stacking. The geometry of the 9-BBN unit leads to a short  $\text{H}91\cdots\text{H}132$  contact ( $1.91\text{ \AA}$ ).

### Experimental

Methoxy-9-BBN (1.50 ml, 1.0 M in hexanes; 1.50 mmol) was added to a solution of bis(diethylglyoximato)nickel(II) (0.10 g, 0.29 mmol) in toluene (5 ml). After heating to reflux for 2 d, the product was chromatographed ( $\text{SiO}_2$ , toluene) to yield (I) (0.114 g, 67%) as orange crystals. M.p. 532–534 K;  $\delta_H$  (400 MHz,  $\text{CDCl}_3$ ): 2.61 (8H, *q*), 1.88–1.80 (4H, *m*), 1.75–1.60 (20H, *m*), 1.50–1.43 (4H, *m*), 1.20, (6H, *t*);  $\delta_C$  (100 MHz,  $\text{CDCl}_3$ ) 160.4, 31.6, 24.8, 19.7, 10.2;  $m/z$  (APCI $^+$ ) 585.56 [ $M+\text{H}^+$ ]. Crystals of (I) were grown from acetone by evaporation.

**Crystal data**[Ni(C<sub>28</sub>H<sub>48</sub>B<sub>2</sub>N<sub>4</sub>O<sub>4</sub>)] $M_r = 585.05$ Monoclinic,  $P2_1/c$  $a = 7.4047(1)$  Å $b = 19.2467(2)$  Å $c = 10.5508(2)$  Å $\beta = 94.7398(5)^\circ$  $V = 1498.52(4)$  Å<sup>3</sup> $Z = 2$ 

$D_x = 1.297$  Mg m<sup>-3</sup>  
 Mo K $\alpha$  radiation  
 Cell parameters from 3483  
 reflections  
 $\theta = 5\text{--}27^\circ$   
 $\mu = 0.69$  mm<sup>-1</sup>  
 $T = 150$  K  
 Block, orange  
 $0.20 \times 0.20 \times 0.20$  mm

**Data collection**

Enraf–Nonius KappaCCD diffractometer

 $\omega$  scans

Absorption correction: multi-scan (*DENZO/SCALEPACK*; Otwinowski & Minor, 1997)  
 $T_{\min} = 0.87$ ,  $T_{\max} = 0.87$   
 3497 measured reflections

**Refinement**Refinement on  $F$  $R = 0.025$  $wR = 0.031$  $S = 1.04$ 

2934 reflections

178 parameters

H-atom parameters not refined

3386 independent reflections  
 2934 reflections with  $I > 3\sigma(I)$   
 $R_{\text{int}} = 0.03$   
 $\theta_{\max} = 27.4^\circ$   
 $h = 0 \rightarrow 9$   
 $k = 0 \rightarrow 24$   
 $l = -13 \rightarrow 13$

Weighting scheme: Prince-modified Chebychev polynomial with 3 parameters (Watkin, 1994),  
 $W = [w][1 - (\Delta F/6\sigma(F))]^2$ ,  
 $w = 2.03, 0.0469$  and  $1.40$   
 $(\Delta/\sigma)_{\max} < 0.001$   
 $\Delta\rho_{\max} = 0.31$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.28$  e Å<sup>-3</sup>

**Table 1**

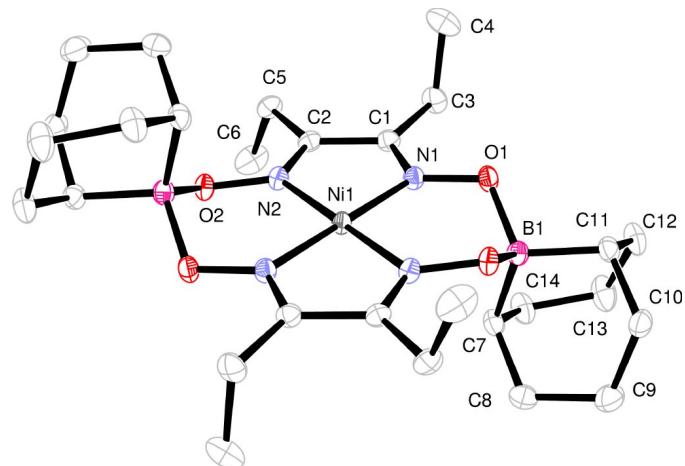
Selected geometric parameters (Å, °).

Ni1—N2	1.8576 (9)	O2—N2	1.3508 (12)
Ni1—N2 <sup>i</sup>	1.8576 (9)	O2—B1 <sup>i</sup>	1.5367 (14)
Ni1—N1	1.864 (1)	N1—C1	1.2984 (15)
Ni1—N1 <sup>i</sup>	1.864 (1)	N2—C2	1.2956 (15)
O1—N1	1.3540 (12)	C1—C2	1.4740 (15)
O1—B1	1.5357 (14)		
Ni1—N1—O1	124.50 (7)	N1—C1—C2	112.1 (1)
Ni1—N1—C1	116.26 (8)	N2—Ni1—N2 <sup>i</sup>	179.994
Ni1—N2—O2	124.72 (7)	N2—Ni1—N1	82.71 (4)
Ni1—N2—C2	116.41 (8)	N2 <sup>i</sup> —Ni1—N1	97.29 (4)
O1—N1—C1	118.99 (9)	N2—Ni1—N1 <sup>i</sup>	97.29 (4)
O2—N2—C2	118.76 (9)	N2 <sup>i</sup> —Ni1—N1 <sup>i</sup>	82.71 (4)
O2 <sup>i</sup> —B1—O1	107.68 (9)	N2—O2—B1 <sup>i</sup>	113.43 (8)
N1—Ni1—N1 <sup>i</sup>	179.994	N2—C2—C1	112.46 (9)
N1—O1—B1	113.41 (8)		

Symmetry code: (i)  $2 - x, -y, 2 - z$ .

H atoms were located in a difference Fourier map and their parameters were not refined.

Data collection: *COLLECT* (Nonius, 1997–2001); cell refinement: *DENZO/SCALEPACK* (Otwinowski & Minor, 1997); data reduc-

**Figure 1**

Plot showing the atomic numbering scheme for (I). Displacement ellipsoids are drawn at the 50% probability level for non-H atoms. Only the contents of the asymmetric unit are labelled. The Ni atom lies on an inversion centre.

tion: *DENZO/SCALEPACK*; program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *CRYSTALS* (Watkin *et al.*, 2001); molecular graphics: *CAMERON* (Watkin *et al.*, 1996); software used to prepare material for publication: *CRYSTALS*.

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